

## Phenylpiperazinylalkylamino Substituted Pyridazinones as Potent $\alpha_1$ Adrenoceptor Antagonists

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QSAR models have been used for designing a series of compounds characterized by a *N*-phenylpiperazinylalkylamino moiety linked to substituted pyridazinones, which have been synthesized. Measurements of the binding affinities of the new compounds toward the  $\alpha_{1a}$ -,  $\alpha_{1b}$ -, and  $\alpha_{1d}$ -AR cloned subtypes as well as the 5-HT<sub>1A</sub> receptor have been done validating, at least in part, the estimations of the theoretical models. This study provides insight into the structure activity relationships of the  $\alpha_1$ -ARs ligands and their  $\alpha_1$ -AR/5-HT<sub>1A</sub> selectivity.

### Introduction

$\alpha_1$ -Adrenergic receptors ( $\alpha_1$ -AR) are members of the superfamily of G protein coupled receptors (GPCR) that transduce signals across the cell membrane. The  $\alpha_1$ -ARs mediate the functional effects of catecholamines such as epinephrine and norepinephrine by coupling to the G $\alpha_q$  that induces the activation of phospholipase C, culminating into the phosphoinositide (PI) hydrolysis.

Molecular biology techniques allowed the identification of cDNAs encoding three  $\alpha_1$ -ARs ( $\alpha_{1a}$ ,  $\alpha_{1b}$ , and  $\alpha_{1d}$ ).<sup>1–5</sup> The three recombinant  $\alpha_1$ -ARs correlate closely with the three  $\alpha_1$ -AR subtypes that were identified in native tissues and which mediate their functional responses ( $\alpha_{1A}$ ,  $\alpha_{1B}$ , and  $\alpha_{1D}$ ).<sup>6–9</sup> The presence of these different  $\alpha_1$ -AR subtypes in blood vessels and other smooth muscles points to the importance of developing selective drugs for receptor classification and characterization as well as for therapeutic effectiveness.

A relevant aspect of chemical research is to predict the behavior of new molecules from their structure, prior to synthesis. It is generally assumed that noncovalent forces control ligand–receptor interactions and that these forces can be described in terms of electrostatic and steric effects, which depend on molecular size and shape. Recently, quantitative structure activity relationship (QSAR) analysis based on ad hoc defined supermolecules and the corresponding size-shape descriptors has been employed for deciphering the molecular features responsible for the affinity and selectivity in a wide ranging series of noncongeneric antagonists of the cloned bovine  $\alpha_{1a}$ -, hamster  $\alpha_{1b}$ -, and rat  $\alpha_{1d}$ -ARs.<sup>10</sup> The QSAR models previously obtained have been herein used to design and estimate the binding affinities of a new molecular series of potential  $\alpha_1$ -AR ligands.

This work describes the design and the synthesis of a series of phenylpiperazine derivatives (**1a–d,k,i,m,o–r**) obtained from the previously re-

ported<sup>11</sup> 2-methyl-4-nitro-5-acetyl-6-substituted-3(2H)-pyridazinones **1a,b**, by nucleophilic substitution at position 4. Several different substituents at position 5 (compounds **1e–h,j,l,n**) have been also considered. The binding affinity assays of the new compounds toward the human cloned  $\alpha_{1a}$ -,  $\alpha_{1b}$ -, and  $\alpha_{1d}$ -AR subtypes as well as the human cloned 5-HT<sub>1A</sub> receptor have been also carried out.

The results of this study provide insight into the structure activity relationships (SAR) of  $\alpha_1$ -ARs ligands and their  $\alpha_1$ -AR/5-HT<sub>1A</sub> selectivity.

### Chemistry

Compounds **1a–d,k,i,m,o–q** were obtained in good yields by nucleophilic displacement of the 4-nitro group of the known pyridazinones **1a,b**,<sup>11</sup> by stirring for 30 min at room temperature an ethanolic solution of the appropriate **1** with an excess of the required *N*, $\omega$ -aminoalkyl-*N*<sup>1</sup>-phenylpiperazine. The latter were in turn prepared by reduction of the corresponding cyanoderivatives or by deprotection of the appropriate phthalimide, following standard procedures. (See Scheme 1). Compound **1r** (see Table 1) was obtained in the same way by directly condensing **1a** and *N*-(2-methoxyphenyl)piperazine. The homologues **1e,f** were prepared according to the same scheme starting from the previously reported 4-nitro-5-propanonyl- (**1c**)<sup>11</sup> and 4-nitro-5-butanonyl (**1d**)<sup>12</sup> analogues. The 5-unsubstituted **1g,h,j,l,n** (see Table 1) were obtained from the appropriate 4-chloropyridazinones,<sup>13,14</sup> by heating at 140 °C with an excess of the required *N*, $\omega$ -aminoalkyl-*N*<sup>1</sup>-phenylpiperazine.

### Results and Discussion

In this work, a theoretical QSAR approach based on size and shape descriptors has been employed to design novel  $\alpha_1$ -AR ligands, taking also advantage of the information inferred from a previous study on monocyclic and bicyclic substituted pyridazinones.<sup>15</sup>

A typical  $\alpha_1$ -blocker contains an unsaturated or aromatic ring on each side of a protonated nitrogen atom, one of these two moieties being closer to the

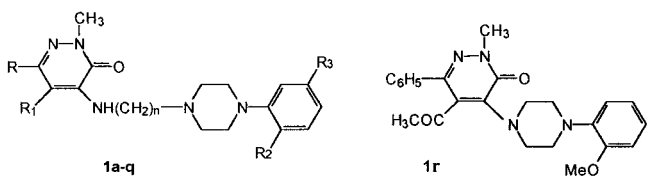
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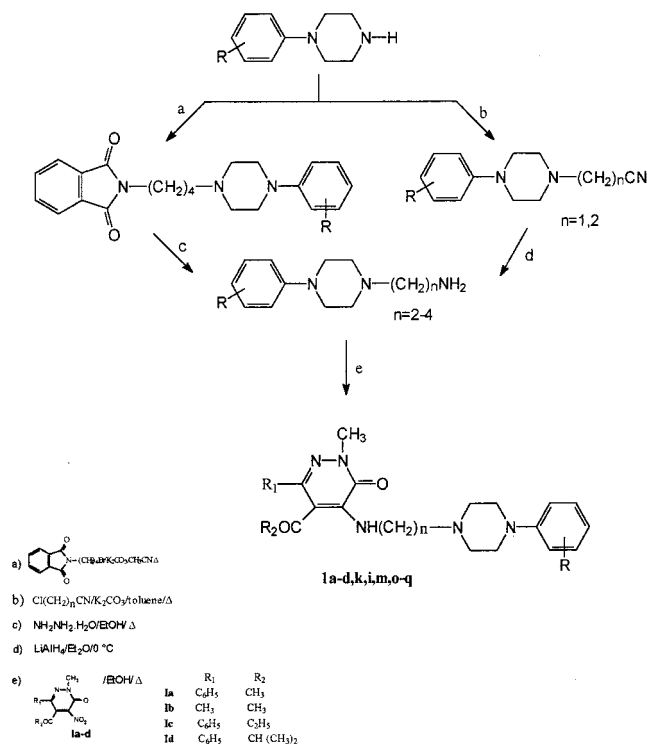
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**Table 1.** Physicochemical Data of Compounds **1a–r**


compd	<i>n</i>	R	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	% yield	mp (°C)	formula <sup>a</sup>
<b>1a</b>	2	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	H	84	oil	C <sub>26</sub> H <sub>30</sub> N <sub>4</sub> O <sub>3</sub>
<b>1b</b>	2	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	F	H	91	147–148	C <sub>25</sub> H <sub>28</sub> FN <sub>5</sub> O <sub>2</sub>
<b>1c</b>	2	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	Cl	H	98	162–163	C <sub>25</sub> H <sub>28</sub> ClN <sub>5</sub> O <sub>2</sub>
<b>1d</b>	2	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	Cl	83	oil	C <sub>26</sub> H <sub>30</sub> ClN <sub>5</sub> O <sub>3</sub>
<b>1e</b>	2	C <sub>6</sub> H <sub>5</sub>	COC <sub>2</sub> H <sub>5</sub>	OCH <sub>3</sub>	H	76	oil	C <sub>27</sub> H <sub>33</sub> N <sub>5</sub> O <sub>3</sub>
<b>1f</b>	2	C <sub>6</sub> H <sub>5</sub>	COCH(CH <sub>3</sub> ) <sub>2</sub>	OCH <sub>3</sub>	H	92	oil	C <sub>28</sub> H <sub>35</sub> N <sub>5</sub> O <sub>3</sub>
<b>1g</b>	2	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>3</sub>	H	53	133–134	C <sub>24</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub>
<b>1h</b>	2	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>3</sub>	Cl	48	oil	C <sub>24</sub> H <sub>28</sub> ClN <sub>5</sub> O <sub>2</sub>
<b>1k</b>	3	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	H	95	oil	C <sub>27</sub> H <sub>33</sub> N <sub>4</sub> O <sub>3</sub>
<b>1i</b>	3	C <sub>6</sub> H <sub>5</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	Cl	72	128–129	C <sub>27</sub> H <sub>32</sub> ClN <sub>5</sub> O <sub>3</sub>
<b>1j</b>	3	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>3</sub>	H	23	130–131	C <sub>25</sub> H <sub>31</sub> N <sub>5</sub> O <sub>2</sub>
<b>1l</b>	3	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>3</sub>	Cl	41	oil	C <sub>25</sub> H <sub>30</sub> ClN <sub>5</sub> O <sub>2</sub>
<b>1m</b>	2	CH <sub>3</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	H	79	133–134	C <sub>21</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub>
<b>1n</b>	2	CH <sub>3</sub>	H	OCH <sub>3</sub>	H	22	177–178	C <sub>19</sub> H <sub>27</sub> N <sub>5</sub> O <sub>2</sub>
<b>1o</b>	3	CH <sub>3</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	H	94	83–84	C <sub>22</sub> H <sub>31</sub> N <sub>5</sub> O <sub>3</sub>
<b>1p</b>	3	CH <sub>3</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	Cl	79	101–102	C <sub>22</sub> H <sub>30</sub> ClN <sub>5</sub> O <sub>2</sub>
<b>1q</b>	4	CH <sub>3</sub>	COCH <sub>3</sub>	OCH <sub>3</sub>	Cl	79	101–102	C <sub>22</sub> H <sub>30</sub> ClN <sub>5</sub> O <sub>3</sub>
<b>1r</b>						95	173–175	C <sub>24</sub> H <sub>26</sub> N <sub>4</sub> O <sub>3</sub>

<sup>a</sup> Elemental analyses for C,H,N were within  $\pm 0.4\%$  of the calculated data.

**Scheme 1**

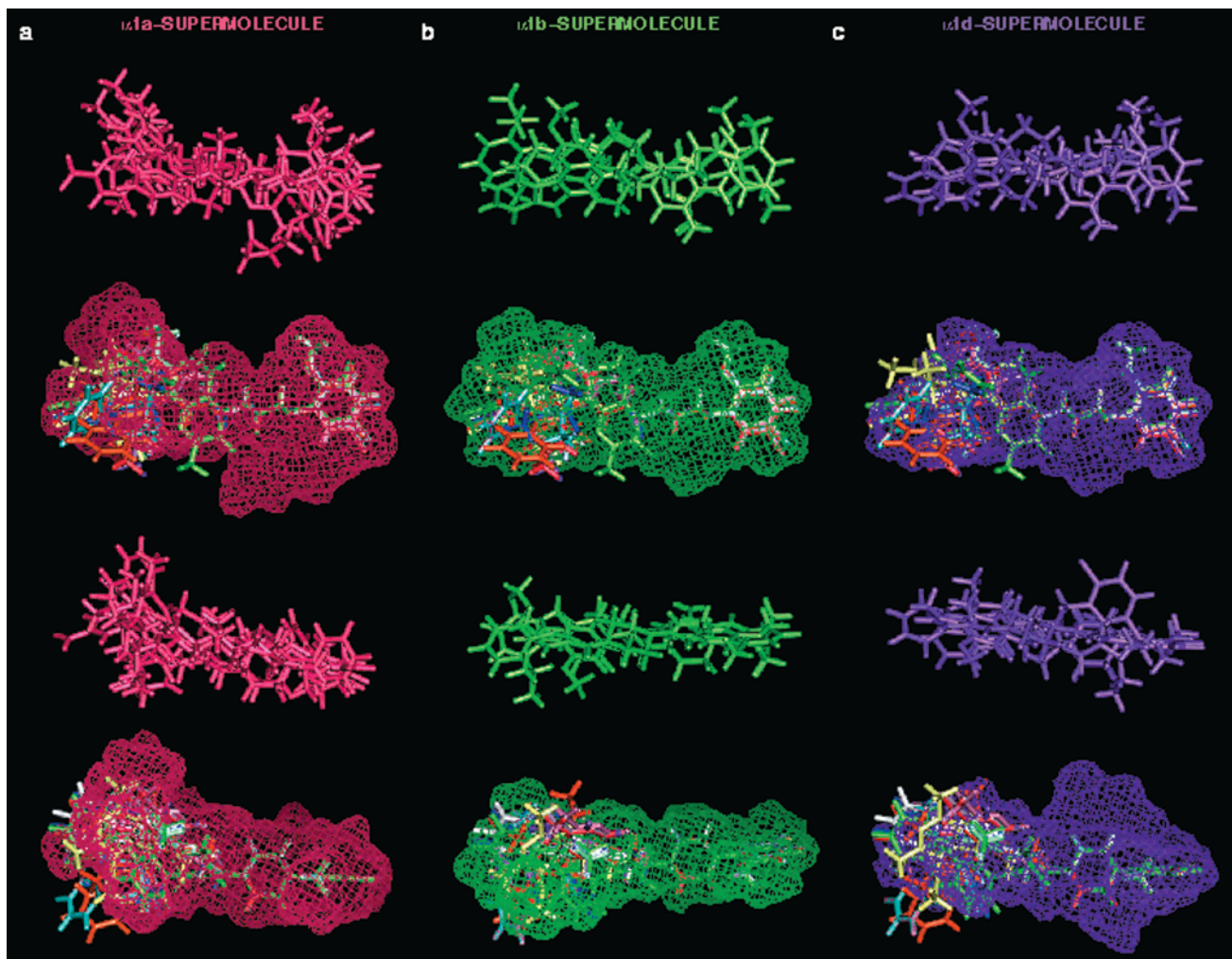
protonated nitrogen atom than the other. The protonated nitrogen atom of the ligands is thought to interact with the conserved aspartate on helix 3 of the receptor according to a precise geometry dictated by a directional charge reinforced hydrogen bonding interaction.<sup>15,16</sup> The protonated nitrogen atom of the  $\alpha_1$ -AR ligands may be found as part of aliphatic chains, of conjugated systems (i.e. quinazoline, isoquinoline, imidazoline) as well as of aliphatic rings (i.e. piperazine and piperidine). While the protonated nitrogen atom is an essential pharma-

cophoric element for the long-range electrostatic recognition and productive interaction with the aspartate of the  $\alpha_1$ -AR binding site, its contribution to the binding energy is almost constant.<sup>16</sup> On the contrary, short-range forces (repulsive, dispersive, inductive, etc.) modulate both ligand affinity and selectivity. Thus, the modulation of the binding affinity by a wide noncongeneric series of  $\alpha_1$ -AR ligands can be described and explained by the variation of the ligand size-shape features that are related to the short range acting forces. These constitute the foundations of the supermolecule-based QSAR approach previously proposed.<sup>10,17,18</sup>

According to the ligand based three-dimensional (3D) pharmacophore, the supermolecule approach assumes that the volume obtained by superimposing the most structurally different ligands that show the highest affinities for the same receptor (supermolecule) might reflect the overall shape and conformational flexibility of the high-affinity receptor binding site. Therefore, size and shape descriptors can be defined ad hoc (that is, on a specific molecular series and in connection with a specific bioactivity) with respect to the supermolecule.

The supermolecule for the  $\alpha_{1a}$ -AR subtype (Figure 1a) is based on six structurally different compounds (**1–6**, Chart 1). Fewer compounds yielded the supermolecules for the  $\alpha_{1b}$ - and the  $\alpha_{1d}$ -AR subtypes (**7–10** for the  $\alpha_{1b}$ -AR and **7, 8, 10–12** for the  $\alpha_{1d}$ -AR; Chart 1, Figure 1b,c); these two supermolecules share three ligands: compounds **7, 8**, and **10** (Chart 1).

The selected size and shape descriptors of the compounds considered in this study are listed in Table 2. In particular, the table reports the inner ( $V_{in1a}$ ,  $V_{in1b}$ , and  $V_{in1d}$ ), the outer ( $V_{out1a}$ ,  $V_{out1b}$ , and  $V_{out1d}$ ), and the difference ( $V_{dif1a}$ ,  $V_{dif1b}$ , and  $V_{dif1d}$ ) van der Waals (vdw) volumes relative to the reference volume of ad hoc defined supermolecules, one for each subtype (Figure 1). According to the definition of these molecular



**Figure 1.** (a) Frontal (top) and side (bottom) views of the supermolecule obtained by superimposing the most active and structurally different antagonists for the cloned bovine  $\alpha_{1a}$ -AR. The same side views have been employed for representing the volume of the  $\alpha_{1a}$  supermolecule embedding all the ligands considered in this study superimposed on the  $\alpha_{1a}$  supermolecule. The same description of point (a) is valid for points (b) and (c) with the difference that they refer, respectively, to the  $\alpha_{1b}$  and  $\alpha_{1d}$  supermolecules. The van der Waals volumes of the  $\alpha_{1a}$ ,  $\alpha_{1b}$ , and  $\alpha_{1d}$  supermolecules are 1016.00 Å<sup>3</sup>, 680.75 Å<sup>3</sup>, and 754.13 Å<sup>3</sup>, respectively.

descriptors, higher affinities are realized by maximizing  $V_{in}$  and minimizing  $V_{out}$ . For its formulation  $V_{dif}$  [ $V_{dif} = (V_{in} - V_{out})/V_{sup}$ ] is a size and shape descriptor that takes into account the information content codified by both  $V_{in}$  and  $V_{out}$  normalized by the volume of the supermolecule ( $V_{sup}$ ). By definition, the ligands constituting the supermolecules show the following singularities:  $V_{mol} = V_{in}$ ,  $V_{out} = 0$ , and  $V_{in}/V_{mol} = 1$ , where  $V_{mol}$  is the vdw molecular volume. The van der Waals volumes of the  $\alpha_{1a}$ ,  $\alpha_{1b}$ , and  $\alpha_{1d}$  supermolecules are 1016.00 Å<sup>3</sup>, 680.75 Å<sup>3</sup>, and 754.13 Å<sup>3</sup>, respectively. In general, a  $V_{dif}$  value close to that of one of the supermolecule components corresponds to high binding affinity value.

As for the  $\alpha_{1a}$ -AR subtype, theoretical size-shape descriptors predict that a two carbon atom linker is compatible with the structural variability of the substituent in position 5 of the pyridazinone ring. In other words, changing the size-shape of the substituent in position 5 of the pyridazinone ring invariably produces high  $V_{dif}$ . In fact, **1a**, **1e**, and **1f** mainly realize relatively high  $V_{dif}$ , by means of high  $V_{in}$  and intermediate  $V_{out}$ , whereas compound **1g** realizes high  $V_{dif}$  through low  $V_{in}$  and  $V_{out}$ .

Reducing the size-shape of the substituent in position 6 (**1m**) or in both positions 5 and 6 (**1n**) slightly lowers

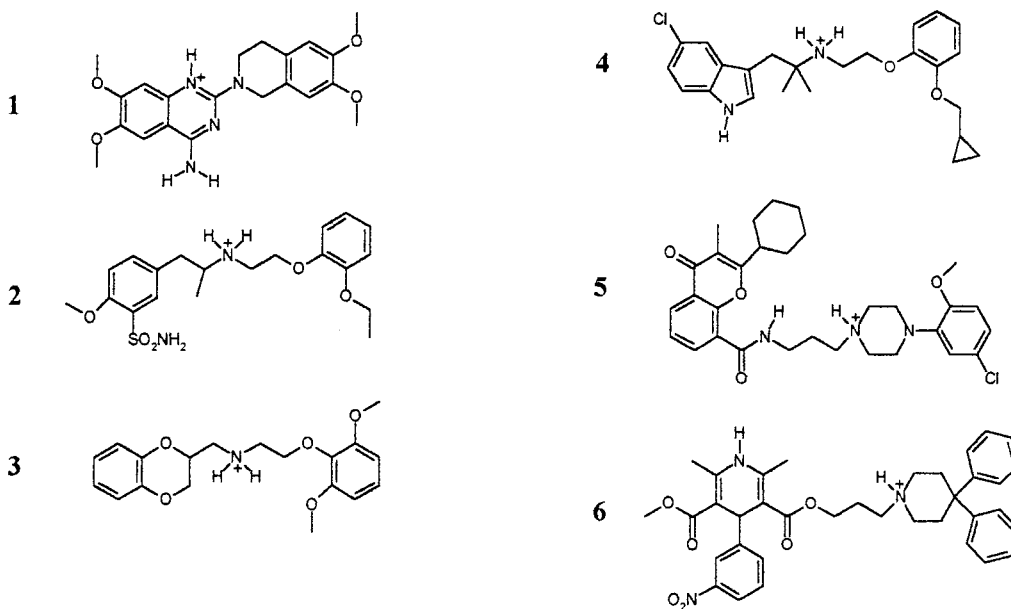
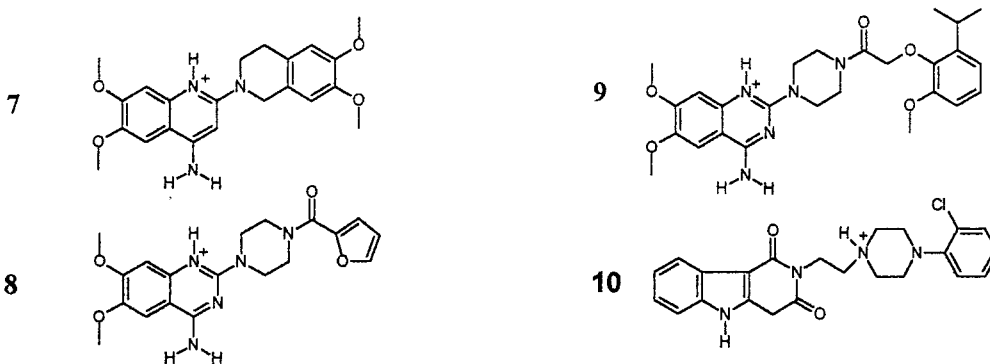
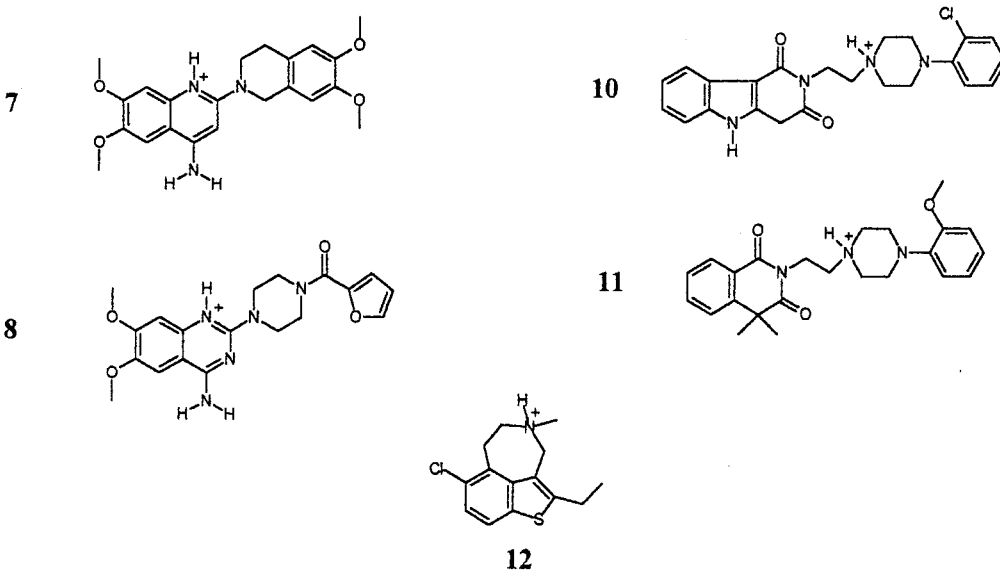
$V_{dif}$  by reducing  $V_{in}$  more than  $V_{out}$ , as compared to compound **1a**.

The fluoro or chloro substitutions for the methoxy group in position 2' of the phenyl ring (**1b** and **1c**) as well as the introduction of a chlorine in position 5' of the phenyl ring (i.e. **1d**) do not induce relevant changes in  $V_{dif}$ , as compared to compound **1a**.

Prolonging the two carbon atom spacer of **1a** to three carbon atoms (**1k**, **1i**), while leaving the phenyl substituent in position 6, increases the outer volume, lowering  $V_{dif}$ . Thus, the computed indices predict that size-shape variability in position 5 of the pyridazinone is well tolerated, whereas increasing the length of the spacer from 2 to 3 methylenic units or substituting the phenyl ring in position 6 of the pyridazinone with a methyl group slightly lowers the  $\alpha_{1a}$ -AR binding affinity.

As for the  $\alpha_{1b}$ -AR, the different substitutions in position 5 of the pyridazinone ring, combined with an ethylenic spacer (**1a**, **1e**, and **1f**), produce relatively high  $V_{dif}$ . Halogen substitutions in positions 2' or 5' of the phenyl ring (**1b-d**) produce  $V_{dif}$  comparable with those of the 2'-methoxy derivative (**1a**). Replacing the phenyl ring in position 6 by a methyl group (**1m**) or reducing the size in both positions 5 and 6 (**1n**) lowers the inner volume more than the outer volume, lowering  $V_{dif}$ .

## Chart 1

Ligands constituting the  $\alpha$ 1a-SUPERMOLECULELigands constituting the  $\alpha$ 1b-SUPERMOLECULELigands constituting the  $\alpha$ 1d-SUPERMOLECULE

**Table 2.** Ad Hoc Defined Size and Shape Descriptors ( $V_{in}$ ,  $V_{out}$ ,  $V_{in}/V_{mol}$ , and  $V_{dif}$ ) of the Set of Molecules Considered in This Study

compd	$V_{in1a}$ ( $\text{\AA}^3$ )	$V_{out1a}$ ( $\text{\AA}^3$ )	$V_{dif1a}$	$V_{in1b}$ ( $\text{\AA}^3$ )	$V_{out1b}$ ( $\text{\AA}^3$ )	$V_{dif1b}$	$V_{in1d}$ ( $\text{\AA}^3$ )	$V_{out1d}$ ( $\text{\AA}^3$ )	$V_{dif1d}$
<b>1a</b>	351.75	45.00	0.3158	350.13	47.62	0.4641	347.75	50.25	0.4124
<b>1b</b>	335.00	43.75	0.2998	334.13	48.50	0.4382	329.00	53.50	0.3819
<b>1c</b>	341.38	45.37	0.3047	343.75	43.88	0.4600	340.63	47.12	0.4069
<b>1d</b>	365.88	46.87	0.3284	358.75	53.13	0.4688	358.00	53.88	0.4216
<b>1e</b>	355.50	56.75	0.3076	349.38	59.87	0.4441	342.38	66.87	0.3819
<b>1f</b>	358.75	66.38	0.3010	356.75	68.25	0.4426	365.50	59.63	0.4240
<b>1g</b>	332.25	31.63	0.3095	327.00	37.13	0.4447	318.00	46.38	0.3765
<b>1h</b>	348.63	33.37	0.3245	336.00	45.88	0.4451	326.38	55.50	0.3755
<b>1k</b>	336.50	73.88	0.2704	307.00	105.00	0.3099	325.38	86.50	0.3311
<b>1i</b>	351.38	74.00	0.2856	314.00	111.75	0.3103	332.38	93.25	0.3315
<b>1j</b>	307.38	69.50	0.2449	296.63	82.87	0.3279	313.38	66.12	0.3428
<b>1l</b>	320.00	73.38	0.2539	297.75	95.13	0.3108	316.00	77.00	0.3313
<b>1m</b>	317.25	26.88	0.2989	309.25	35.63	0.4197	318.88	26.00	0.4060
<b>1n</b>	295.88	14.00	0.2902	283.88	27.75	0.3929	287.25	24.38	0.3644
<b>1o</b>	333.63	25.50	0.3172	306.63	55.87	0.3847	322.13	40.25	0.3908
<b>1p</b>	347.13	27.87	0.3287	324.13	51.62	0.4180	333.75	41.75	0.4048
<b>1q</b>	358.00	29.88	0.3378	317.63	70.87	0.3785	293.13	95.00	0.2747
<b>1r</b>	302.88	52.25	0.2580	296.75	60.88	0.3618	304.75	52.63	0.3495

**Table 3.** Calculated by  $V_{dif}$  Based QSAR Models ( $pK_{1a\text{ calc}}$ ,  $pK_{1b\text{ calc}}$ ,  $pK_{1d\text{ calc}}$ ) and Experimental ( $pK_{1a\text{ exp}}$ ,  $pK_{1b\text{ exp}}$ ,  $pK_{1d\text{ exp}}$ ) Binding Affinities

compd	$pK_{1a\text{ calc}}$	$pK_{1a\text{ exp}}$	$\Delta$	$pK_{1b\text{ calc}}$	$pK_{1b\text{ exp}}$	$\Delta$	$pK_{1d\text{ calc}}$	$pK_{1d\text{ exp}}$	$\Delta$
<b>1a</b>	8.82	9.46	-0.64	8.98	8.48	0.50	9.19	9.89	-0.70
<b>1b</b>	8.74	9.49	-0.75	8.83	8.42	0.41	8.89	9.31	-0.42
<b>1c</b>	8.77	9.52	-0.75	8.96	9.12	-0.16	9.13	9.60	-0.47
<b>1d</b>	8.89	9.55	-0.66	9.01	8.40	0.61	9.28	9.39	-0.11
<b>1e</b>	8.78	9.72	-0.94	8.87	8.84	0.03	8.89	9.60	-0.71
<b>1f</b>	8.75	9.74	-0.99	8.86	9.19	-0.33	9.30	9.37	-0.07
<b>1g</b>	8.79	9.66	-0.87	8.87	8.70	0.17	8.84	7.93	0.91
<b>1h</b>	8.87	9.00	-0.13	8.87	8.75	0.12	8.83	8.97	-0.14
<b>1k</b>	8.59	8.82	-0.23	8.08	7.69	0.39	8.39	8.37	0.02
<b>1i</b>	8.67	8.64	0.03	8.08	8.85	-0.77	8.40	8.41	-0.01
<b>1j</b>	8.46	8.59	-0.13	8.18	8.26	-0.08	8.51	7.93	0.58
<b>1l</b>	8.50	8.70	-0.20	8.04	8.70	-0.66	8.40	8.53	-0.13
<b>1m</b>	8.74	8.64	0.10	8.72	7.30	1.42	9.12	8.50	0.62
<b>1n</b>	8.69	7.73	0.96	8.57	7.48	1.09	8.72	7.75	0.97
<b>1o</b>	8.83	9.09	-0.26	8.52	7.27	1.25	8.98	7.46	1.52
<b>1p</b>	8.89	9.29	-0.40	8.71	8.10	0.61	9.11	8.27	0.84
<b>1q</b>	8.94	8.52	0.42	8.48	7.39	1.09	7.84	8.30	-0.46
<b>1r</b>	8.53	6.79	1.74	8.38	6.59	1.79	8.57	5.00	3.57

<sup>a</sup>  $pK_{1a}$  values have been calculated according to the following regression equation:  $pK_{1a} = 5.18 (\pm 0.83) V_{dif} + 7.19 (\pm 0.27)$ ,  $n = 34$ ,  $r = 0.74$ ,  $s = 0.53$ , and  $F = 39.01$ , where  $n$  is the number of compounds,  $r$  is the correlation coefficient,  $s$  is the saturated deviation, and  $F$  is the Fisher ratio. <sup>b</sup>  $pK_{1b}$  values have been calculated according to the following regression equation:  $pK_{1b} = 5.87 (\pm 0.72) V_{dif} + 6.26 (\pm 0.22)$ ,  $n = 30$ ,  $r = 0.83$ ,  $s = 0.51$ , and  $F = 66.08$ . <sup>c</sup>  $pK_{1d}$  values have been calculated according to the following regression equation:  $pK_{1d} = 0.74 (\pm 1.14) V_{dif} + 5.17 (\pm 0.33)$ ,  $n = 31$ ,  $r = 0.84$ ,  $s = 0.57$ , and  $F = 72.86$ .

Moreover, the elongation of the ethylenic spacer of **1a** to three carbon atoms (**1k**) reduces  $V_{in}$  and increases  $V_{out}$ , lowering  $V_{dif}$ . In general, compounds with a three or four carbon atom spacer show lower  $V_{dif}$  than the corresponding two carbon atom spacer compounds (Tables 1–4).

Thus, for the  $\alpha_{1b}$ -AR, size-shape indices predict that, similarly to the  $\alpha_{1a}$ -AR, structural variability in position 5 of the pyridazinone is better tolerated than that in position 6. Moreover, the elongation of the alkyl spacer is predicted to lower the  $\alpha_{1b}$ -AR binding affinity.

As for the  $\alpha_{1d}$ -AR, size and shape descriptors show a behavior almost similar to that of the  $\alpha_{1a}$ -AR.

The linear regression equations previously obtained (Table 3)<sup>10</sup> have been used for estimating the binding affinities of the new compounds. On the basis of both the predicted binding affinities data values (nanomolar range) and the smaller variance of the estimated  $\alpha_{1a}$ -AR binding affinities (about 0.5 in a logarithmic scale) with respect to the other two subtypes (about 1.0 and 1.5, respectively), we decided to synthesize the designed compounds.

The predicted and experimental binding affinities for the three  $\alpha_1$ -AR subtypes, together with their differences ( $\Delta$ ), are reported in Table 3. Recalling that the simple approach used is mainly devoted to ligand design purposes and is based on QSAR models generated by a highly noncongeneric molecular series,<sup>10</sup> the agreement between the predicted and measured binding affinity data values can be considered satisfactory. In fact, we consider these QSAR models as useful ligand design and decision making tools. In this case, these tools showed low predictive resolution (about 0.5, 1, and 1.5 respectively, in a logarithmic scale). Nevertheless, they allowed the selection among the many hypothetical ligands designed by estimating their affinities for the  $\alpha_1$ -ARs according to three levels: (a) low affinity,  $pK_i$  7–8; (b) high affinity,  $pK_i$  8–9; and (c) very high affinity,  $pK_i > 9$ . On these bases, a few wrong estimations ( $\Delta > 1.1$ ) have been done ( $pK_{1a}$ : compound **1r**;  $pK_{1b}$ : compounds **1m**, **1o**, **1r**;  $pK_{1d}$ : compounds **1o** and **1r**). The molecular descriptor  $V_{dif}$  completely failed in estimating the affinities of **1r** for all the three  $\alpha_1$ -AR subtypes, suggesting that the low binding affinity of **1r** does not depend on

**Table 4.** Affinity Constants ( $K_i$ , nM) of Compounds **1a–r**; Prazosin, **BMY7378**, and **8-OH-DPAT** toward Human Cloned  $\alpha_1$  Adrenoceptor Subtypes and 5-HT<sub>1A</sub> Receptor<sup>a</sup>

compd	$K_i$ (nM) <sup>a</sup>				binding <sup>b</sup> [ <sup>35</sup> S] GTP $\gamma$ S
	$\alpha$ -1a	$\alpha$ -1b	$\alpha$ -1d	5-HT <sub>1A</sub>	
<b>1a</b>	0.63	3.29	0.13	2.47	n.a.
<b>1b</b>	0.32	3.77	0.49	24.12	
<b>1c</b>	0.30	0.75	0.25	11.60	
<b>1d</b>	0.28	3.95	0.41	76.23	
<b>1e</b>	0.19	1.45	0.25	4.06	pEC <sub>50</sub> = 8.6; 19%
<b>1f</b>	0.18	0.64	0.43	10.20	n.a.
<b>1g</b>	0.22	2.01	11.78	13.98	
<b>1h</b>	1.0	1.76	1.07	322.1	
<b>1k</b>	1.51	20.28	4.32	134.50	
<b>1i</b>	0.44	1.44	3.92	1000.00	
<b>1j</b>	2.59	5.51	11.79	126.72	
<b>1l</b>	2.0	2.01	2.92	> 1000	
<b>1m</b>	2.29	49.84	3.20	13.00	
<b>1n</b>	18.56	33.13	17.69	32.10	
<b>1o</b>	0.89	53.39	34.45	1.48	pEC <sub>50</sub> = 8.8; 17%
<b>1p</b>	0.53	8.04	5.34	11.24	
<b>1q</b>	2.99	49.08	5.96	1.39	
<b>1r</b>	162.80	258.92	> 10.000	> 1000	
<b>Prazosin</b>	0.61	0.42	0.23	> 10.000	
<b>BMY7378</b>	381.05	68.97	1.43	0.93	pEC <sub>50</sub> = 9.27; 26%
<b>8-OH-DPAT</b>	1757.20	5975.00	> 1000	3.44	pEC <sub>50</sub> = 76; 100%

<sup>a</sup> Equilibrium dissociation constants ( $K_i$ ) were derived from IC<sub>50</sub> using the Cheng-Prusoff equation.<sup>23</sup> The affinities estimated were derived from displacement of [<sup>3</sup>H]prazosin binding for  $\alpha_1$ -adrenoceptors and [<sup>3</sup>H]-8-hydroxy-2-(di-*n*-propylamino)tetraline for 5-HT<sub>1A</sub> receptor. Each experiment was performed in triplicate.  $K_i$  values were from 2 to 3 experiments which agreed within 20%. <sup>b</sup> n.a. = not active, indicative of neutral antagonism. For partial agonists the pEC<sub>50</sub> and percent of maximum effect are given. Values are from 2 to 3 experiments which agreed within 20%.

the ligand size-shape features. In fact, the very low affinity of this compound for all the three  $\alpha_1$ -ARs is probably due to the fact that no nitrogen atom suitable for protonation, at physiological pH, is present in this compound, thus preventing the ligand to perform the proper salt-bridge interaction with the aspartate of the receptor binding site. The overestimation of the  $\alpha_{1b}$ -AR binding affinity of **1m** is due to the fact that reducing the size of the pyridazinone moiety, with respect to **1a**, lowers both the inner and outer volumes of this compound with respect to the reference  $\alpha_{1b}$ -AR supermolecule but not enough for producing a good estimation of the binding affinity of this compound. The overestimation of the  $\alpha_{1b}$ - and the  $\alpha_{1d}$ -AR binding affinity of **1o** is due to the fact that reducing the size of the pyridazinone moiety increases  $V_{in}$  and reduces  $V_{out}$  with respect to **1l**, thus producing a  $V_{dif}$  too high for predicting correctly the binding affinity of this compound. As a consequence, QSAR models fail in estimating the  $\alpha_{1a}$ -AR selectivity of **1o**.

In general,  $\alpha_{1a}$ -AR binding affinities are slightly underestimated (Table 3) whereas, the binding affinities for the other two  $\alpha_1$ -AR subtypes, in particular the  $\alpha_{1b}$ -AR, are rather overestimated (Table 3) by the theoretical model. For these reasons, the theoretical model does not estimate properly the  $\alpha_{1a}/\alpha_{1b}$  selectivity of compounds **1m**, **1o**, **1p**, and **1q** and the  $\alpha_{1a}/\alpha_{1d}$  selectivity of compounds **1g**, **1o**, and **1p**.

SAR analysis shows that three compounds, i.e., **1g**, **1o**, and **1p**, are selective for the  $\alpha_{1a}$ -AR as compared to the other two  $\alpha_1$ -AR subtypes. Selectivity of **1g** is mainly due to the lack of substitution in position 5 of the pyridazinone that primarily lowers the affinity for the  $\alpha_{1d}$ -AR and, by a lower extent, that for the  $\alpha_{1b}$ -AR. The  $\alpha_{1a}$ -AR selectivity of compound **1o** is mainly due to the replacement of the phenyl ring in position 6 of the pyridazinone with a methyl group, on one hand, that lowers the affinity for the  $\alpha_{1b}$ -AR, and to the elongation

of the ethylenic spacer to three carbon atoms, on the other one, that lowers the affinity for the  $\alpha_{1d}$ -AR. However, the presence of a phenyl ring in position 6 increases the affinity of this compound for the 5-HT<sub>1A</sub> receptor. This drawback is overcome by introducing a chlorine in position 5' of the phenyl ring (i.e. **1p**), that lowers the affinity for the serotonergic receptor while also retaining lower affinities for the  $\alpha_{1b}$ - and  $\alpha_{1d}$ -AR subtypes, than for the  $\alpha_{1a}$ -AR.

Interestingly, the majority of the compounds proposed in this work show low affinities for the 5-HT<sub>1A</sub> receptor. This is mainly due to the presence of a chlorine substituent in position 5' of the phenyl ring and/or to the elongation of the alkyl spacer. However, the latter needs the presence of a phenyl ring in position 6 of the pyridazinone to produce a low affinity for the 5-HT<sub>1A</sub> receptor.

As for functional studies, compound **1i** was tested for its functional activity on rat aorta contractions induced by (-)-noradrenaline and, as expected, proved to be an antagonist with pK<sub>b</sub> of 8.21.

Furthermore, compounds showing a relevant affinity for the 5-HT<sub>1A</sub> receptor were tested for their ability to influence the binding of [<sup>35</sup>S]GTP $\gamma$ S to HeLa cell membranes expressing the human 5-HT<sub>1A</sub> receptor. Compounds **1a**, **1f**, and **1q** behaved as neutral antagonists, whereas compounds **1e** and **1o** proved to be partial agonists with a pEC<sub>50</sub> of 8.6 and 8.8 and a percent maximum effect of 19 and 17%, respectively. (See Table 4.)

In conclusion, the QSAR models previously obtained on a very heterogeneous series of protonated  $\alpha_1$ -AR antagonists have been challenged in their capability to design and estimate the  $\alpha_1$ -AR binding affinities of a set of *N*-arylpiperazines with satisfactory results.

This study indicates that substitutions in positions 5 and/or 6 of the pyridazinone ring as well as in positions 2' and 5' of the phenyl ring together with the length of

the linker are responsible for modulating the ligand  $\alpha_1$ -AR binding affinities and  $\alpha_1$ -AR/5-HT<sub>1A</sub> selectivities.

## Experimental Section

**Chemistry.** Melting points were determined on a Büchi 510 capillary melting points apparatus and are uncorrected. Analyses indicated by the symbols were within  $\pm 0.4$  of the theoretical values. <sup>1</sup>H NMR spectra were recorded on a Bruker AC200 spectrometer; chemical shifts are reported as  $\delta$  (ppm) relative to tetramethylsilane. TLC on silica gel plates was used to check product purity. Silica gel 60 (Merck; 230–400 mesh) was used for flash chromatography.

**2-Methyl-4-[(4-arylpiperazin-1-yl)aminoalkyl]-5-acetyl-6-phenyl(methyl)-3-(2H)-pyridazinones 1a–d,k,i,m,o–q.** A solution of the required **1a,b**<sup>11</sup> (1 mmol) and the appropriate *N*, $\omega$ -aminoalkyl-*N*<sup>1</sup>-phenylpiperazine<sup>15</sup> (2.5 mmol) in EtOH (6 mL) was stirred at room temperature for 30 min. Water (5 mL) was added, the solvent was evaporated under vacuum, and the residue was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, the solid was filtered off, and the solvent was evaporated under vacuum. The residue was purified by flash chromatography eluting with cyclohexane/ethyl acetate 4/6 to give as the first run the desired compounds **1**. (See Table 1 for data.)

**For 1a:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.9 (s, 3H); 2.7–2.8 (m, 6H); 3.1–3.2 (m, 4H); 3.3–3.4 (m, 2H); 3.8 (s, 3H); 3.9 (s, 3H); 6.8–7.0 (m, 4H); 7.4 (s, 5H); 8.0 (br. s, 1H, exch. with D<sub>2</sub>O).

The higher homologues **1e,f** were prepared according to the same method starting from 2-methyl-4-nitro-5-propanonyl-**(1c)**<sup>11</sup> or isobutanonyl-**(1d)**,<sup>12</sup> respectively.

**For 1e:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 0.8 (t, 3H, *J* = 8 Hz); 2.1 (q, 2H; *J* = 8 Hz); 2.6–2.8 (m, 6H); 3.0–3.2 (m, 6H); 3.8 (s, 3H); 3.9 (s, 3H); 6.8–7.0 (m, 4H); 7.3–7.5 (m, 6H)

**For 1f:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 0.8 (2s, 6H); 2.2–2.3 (m 1H); 2.6–2.7 (m, 6H); 3.0–3.1 (m, 4H); 3.2–3.6 (m, 2H); 3.8 (s, 3H); 3.9 (s, 3H); 6.8–7.1 (m, 4H); 7.3–7.5 (m, 6H).

Compound **1r** was obtained by condensation in the same conditions of **1a** and *N*-(2-methoxyphenyl)piperazine. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 2.1 (s, 3H); 3.1–3.2 (m, 4H); 3.5–3.6 (m, 4H); 3.9 (2s, 6H); 6.9–7.0 (m, 4H); 7.4 (app.s, 5H). (See Table 1 for data.)

**2-Methyl-4-[(4-arylpiperazin-1-yl)aminoalkyl]-6-phenyl(methyl)-3-(2H)-pyridazinones 1g,h,j,l,n.** A mixture of the required 4-chloro-3-(2H)pyridazinone<sup>13,14</sup> and an excess of the appropriate phenylpiperazinylamine was heated at 140 °C for 12 h. The residue was first purified by flash chromatography eluting with CH<sub>2</sub>Cl<sub>2</sub>/MeOH 95/5 and then by gravimetric chromatography, eluting with toluene/methanol 9/1. (See Table 1 for data.)

**Biology. Radioligand Binding Assay at Cloned  $\alpha_1$ -Adrenoceptors.** Binding to cloned human  $\alpha_1$ -adrenoceptor subtypes was performed in membranes from CHO cells (chinese hamster ovary cells) transfected by electroporation with DNA expressing the gene encoding each  $\alpha_1$ -adrenoceptor subtype. Cloning and stable expression of the human  $\alpha_1$ -adrenoceptor gene was performed as previously described.<sup>19</sup> CHO cell membranes (30  $\mu$ g proteins) were incubated in 50 mM Tris-HCl, pH 7.4, with 0.1–0.4 nM [<sup>3</sup>H]prazosin, in a final volume of 1 mL for 30 min at 25 °C, in the absence or presence of competing drugs (1 pM–10  $\mu$ M). Nonspecific binding was determined in the presence of 10  $\mu$ M phentolamine. The incubation was stopped by addition of ice-cold Tris-HCl buffer and rapid filtration through 0.2% polyethyleneimine pretreated Whatman GF/B or Schleicher and Schuell GF52 filters.

**Radioligand Binding Assay at Human Cloned 5HT<sub>1A</sub>-Serotonergic Receptors.** Genomic clone G-21 coding for the human 5HT<sub>1A</sub>-serotonergic receptor was stably transfected in a human cell line (HeLa).<sup>20</sup> HeLa cells were grown as monolayers in Dulbecco's modified Eagle's medium (DMEM), supplemented with 10% fetal calf serum and gentamicin (100  $\mu$ g/mL), 5% CO<sub>2</sub> at 37 °C. Cells were detached from the growth flask at 95% confluence by a cell scraper and were lysed in ice-cold Tris 5 mM and EDTA 5 mM buffer (pH 7.4). Homogenates were centrifuged at 40 000  $\times$  g  $\times$  20 min, and pellets

were resuspended in a small volume of ice-cold Tris 5 mM and EDTA 5 mM buffer (pH 7.4) and immediately frozen and stored at –70 °C until use.

On the experimental day, cell membranes were resuspended in binding buffer 50 mM Tris (pH 7.4), 2.5 mM MgCl<sub>2</sub>, 10  $\mu$ M pargiline.<sup>21</sup> Membranes were incubated in a final volume of 1 mL for 30 min at 30 °C with 1.2 nM [<sup>3</sup>H]8-OH-DPAT, in the absence or presence of competing drugs; nonspecific binding was determined in the presence of 10  $\mu$ M 5-HT. The incubation was stopped by addition of ice-cold Tris buffer and rapid filtration through 0.2% polyethyleneimine pretreated Schleicher and Schuell GF52 filters.

The inhibition of specific binding of the radioligands by the tested drugs was analyzed to estimate the IC<sub>50</sub> value by using the nonlinear curve-fitting program Allfit.<sup>22</sup> The IC<sub>50</sub> value is converted to an affinity constant (*K*<sub>i</sub>) by the equation of Cheng and Prusoff.<sup>23</sup>

**Functional Studies.** Evaluation of compounds for  $\alpha_1$ -antagonism was performed as previously published.<sup>24</sup> Briefly, Sprague Dawley rats (350–450 g body weight) were sacrificed by cervical dislocation. The aorta was isolated, freed of adhering connective tissue, and placed in Krebs solution containing NaCl (112 mM), glucose (11.1 mM), NaHCO<sub>3</sub> (25 mM), KCl (4.7 mM), CaCl<sub>2</sub> (2.5 mM), KH<sub>2</sub>PO<sub>4</sub> (1.2 mM), and MgSO<sub>4</sub> (1.2 mM). Desmethylimipramine (0.1  $\mu$ M) and corticosterone (1  $\mu$ M) to block neuronal and extraneuronal uptake of noradrenaline, ( $\pm$ )-propranolol (1  $\mu$ M) to block  $\beta$ -receptors, and yohimbine (0.1  $\mu$ M) to block  $\alpha_2$ -receptors were added to the Krebs solution. Aortic strips (2  $\times$  30 mm long) were mounted for isotonic tension recording in 20 mL-organ bath containing Krebs buffer aerated constantly with 95% O<sub>2</sub>–5% CO<sub>2</sub> and maintained at 37 °C and loaded with a resting tension of 1.5 g. The strips were allowed to equilibrate for 60 min with washing every 20 min. After the equilibration period, tissues were primed twice (every 60 min) by addition of 10  $\mu$ M noradrenaline. After another washing and equilibration period of 60 min, a noradrenaline concentration–response curve was constructed (basal curve). Following washout of noradrenaline, single concentrations of the compounds were incubated for 30 min before repeating the noradrenaline concentration–response curve. Responses were expressed as percentage of the maximal contraction observed in the basal noradrenaline concentration–response curve and analyzed by nonlinear curve fitting according to the method reported by De Lean et al.<sup>22</sup> Schild-plot parameters were evaluated by linear-regression analysis according to Tallarida and Murray.<sup>25</sup>

**Stimulation of [<sup>35</sup>S]GTP $\gamma$ S Binding at Cloned 5-HT<sub>1A</sub> Receptor.** The effects of the compounds tested with [<sup>35</sup>S]-GTP $\gamma$ S binding were evaluated according to the method of Stanton and Beer<sup>26</sup> with minor modifications. On the experimental day, cell membranes from HeLa cells transfected with human cloned 5-HT<sub>1A</sub> receptors, prepared as above-described, were resuspended in buffer containing 20 mM HEPES, 3 mM MgSO<sub>4</sub>, and 120 mM NaCl (pH 7.4). The membranes were incubated with 30  $\mu$ M GDP and decreasing concentrations of test drugs (from 100  $\mu$ M to 0.1 nM) or decreasing concentrations of 5-HT, from 100  $\mu$ M to 0.1 nM (reference curve) for 20 min at 30 °C in a final volume of about 0.5 mL. Samples were then transferred to ice, added with [<sup>35</sup>S]GTP $\gamma$ S (150–250 pM), and then incubated for a further 30 min at 30 °C. Nonspecific binding was determined in the presence of 10  $\mu$ M GTP $\gamma$ S. The incubation was stopped by addition of ice-cold HEPES and rapid filtration on Schleicher and Schuell GF52 filters, using a Brandel cell harvester. The filters were washed three times with a total of 5 mL of the same buffer. Radioactivity was counted by liquid scintillation spectrometry with efficiency > 90.

Stimulation of [<sup>35</sup>S]GTP $\gamma$ S binding induced by the compounds tested was expressed as percent increase in binding above basal value, being the maximal stimulation observed with 5-HT taken as 100%. The concentration–response curve of the agonistic activity was analyzed by nonlinear fitting program Allfit.<sup>22</sup> The maximal stimulation of [<sup>35</sup>S]GTP $\gamma$ S

binding ( $E_{\max}$ ) achieved for each drug and the concentration required to obtain 50% of  $E_{\max}$  ( $pEC_{50}$  value) were evaluated.

**Molecular Modeling.** The protonated structures of the ligands considered in this study were fully optimized by means of semiempirical molecular orbital calculations (AM1),<sup>27</sup> using the MOPAC 6.0 (QCPE 455) program.

The ad hoc modeling consisted of comparing the vdw volume of the minimized structure of each ligand (in its extended conformation) with the vdw volume of a supermolecule chosen as a template. Three different supermolecules were modeled for the three  $\alpha_1$ -AR subtypes. In the series of structurally heterogeneous ligands considered in a previous study,<sup>10</sup> subsets of analogues can be identified. For each subset, the ligand showing the highest affinity for the specific  $\alpha_1$ -AR subtype was chosen as a component of the respective reference supermolecule. Thus, the ligands used for the  $\alpha_{1a}$  supermolecule are compounds 1–6, the ligands for the  $\alpha_{1b}$  supermolecule are compounds 7–10, and the ligands for the  $\alpha_{1d}$  supermolecule are 7, 8, 10–12 (Chart 1). The ligands chosen as components of each supermolecule were superimposed by a topologic rigid body fit procedure based on the following pharmacophoric criteria: (a) the hydrogen of the protonated nitrogen atom and (b) the aromatic rings closest to and farthest from the protonated nitrogen. All the other compounds were rigidly superimposed on the appropriate supermolecule with each ligand being superimposed on the analogue compound present in the supermolecule or on its structurally closest compound. Matching involved the moieties carrying the protonated nitrogen, i.e., the piperazinic ring.

QUANTA molecular modeling package (release 96; Molecular Simulation Inc., 200 Fifth Avenue, Waltham, MA 02154) was used for molecular comparison and computation of the vdw volumes. We considered the following size and shape descriptors:  $V_{in}$  and  $V_{out}$ , which are, respectively, the intersection and the outer van der Waals volume of the ligand considered with respect to the vdw volume of the reference supermolecule; and  $V_{dif}$ , which is computed according to the formula  $V_{dif} = (V_{in} - V_{out})/V_{sup}$ , where  $V_{sup}$  is the molecular volume of the reference supermolecule.

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